

Corrigenda

Carbohydr. Res., 111 (1982) 163–169.

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line 15: add $\text{S}(\text{CH}_2)_2\text{Ar}$ after OMe.

line 17: delete $\text{S}(\text{CH}_2)_2\text{Ar}$.

Table I

Compound 1 X = $\text{S}(\text{CH}_2)_2\text{Ph}$: $[\text{M}]_{\text{D}}$ should read 449.

Compound 2 X = $\text{OC}_6\text{H}_4\text{NH}_2(p)$ should read $\text{OC}_6\text{H}_4\text{NMe}_2(p)$ $[\text{M}]_{\text{D}}$ 380.

X = $\text{S}(\text{CH}_2)_2\text{Ph}$: $[\text{M}]_{\text{D}}$ should read 594.

X = $\text{S}(\text{CH}_2)_6\text{NH}_2$ should read $\text{S}(\text{CH}_2)_6\text{NHCO}_2\text{CH}_2\text{Ph}$
 $[\text{M}]_{\text{D}}$ 506.

X = $\text{SC}_6\text{H}_4\text{NO}_2(p)$: ref. no. should be 17.

X = $\text{SCH}_2\text{C}_6\text{H}_4\text{NO}_2(p)$: $[\text{M}]_{\text{D}}$ should read 1125; and delete
footnote *f*.

Footnote *d* should read: The sign of the optical rotation is surprisingly negative.

Table III, recalculated values,

Compound 1: $r = 0.940$, $m = 55.3$, $I = 25.8$.

Compound 2: $n = 36$, $r = 0.962$, $m = 92.4$, $I = -203$.

Delete footnote *e*.

Table IV

entry 5: -130 should read -121 .

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lines 6 and 7: delete “or 2” to read ($n = 1$).